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Heat & Its Measurement

Answers

Heat & Its
Measurement 1. How
many Joules of heat are
given off when 5.00
grams of water cool
from 75 °C to 25 °C? Q
 $= m \cdot C_p \cdot \Delta T$ $Q = 5.00 \text{ g}$
 $\cdot 4.18 \text{ J/g}\cdot\text{°C} \cdot (75 - 25$
 $\text{°C}) = 1045 \text{ J}$ 2. If 575
Joules of heat are
added to 45.0 grams of
water at 25.0 °C, what
will the new
temperature be? $Q =$

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Heat & Its Measurement - Dr. Hall's Science Site

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Measurement 1. How many Joules of heat are given off when 5.00 grams of water cool from 75 °C to 25 °C? $Q = m \cdot C_p \cdot \Delta T$
 $Q = 5.00 \text{ g} \cdot 4.18 \text{ J/g}\cdot\text{°C} \cdot (75 - 25 \text{ °C}) = 1045 \text{ J}$

2. If 575 Joules of heat are

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added to 45.0 grams of water at 25.0 °C, what will the new temperature be? $Q = m \cdot$

Heat And Its Measurement Answers

Explain the terms heat and temperature. Heat is a form of energy that flows from a body of higher temperature to one of lower temperature.

Temperature is a

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degree of hotness or coldness of a body. 2. What is thermometry? The science of temperature and its measurement is called thermometry. 3. Discuss various scales of thermometer.

Heat And Temperature Grade 11 Physics Question Answer ...

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HEAT AND ITS
MEASUREMENT Name _____
_____ ~at (or energy)
can be measured in
units of calories or
joules. When there is a
,,,,,nperature change
(AT), heat (Q) can be
calculated using this
formula: $Q = \text{mass} \times L \backslash$
T x specific heat

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capacity ($Q = m \times C \times \Delta T$)
Final Temp - Initial Temp)

During a phase change, we use this formula:
 $Q = m \times L$
mass x heat of fusion (or heat of vaporization) |

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Heat And Its Measurement Answer Key

Answer: Its expansion is uniform. (For equal amounts of heat it expands by equal lengths.) It is opaque and shining. It does not stick to the sides of the glass tube. It is a good

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conductor of heat. It has a high boiling point (357°C) and a low freezing point (-39°C). Hence a wide range of temperatures can be measured using a mercury thermometer.

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Science Chapter 5 ...

A calorimeter is a body in the surroundings of the system, with its own temperature and internal energy; when it is connected to the system by a path for heat transfer, changes in it measure heat transfer. The mechanical view was pioneered by Helmholtz and developed and used in the twentieth century,

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largely through the
influence of Max Born.

Heat - Wikipedia

Heat Class 7 Extra
Questions Science
Chapter 4 Heat Class 7
Science Extra
Questions Very Short
Answer Type. Question
1. In a mercury
thermometer, the level
of mercury rises when
its bulb comes in
contact with a hot
object.

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The answer is (a) is a better conductor of heat than the wooden tile. 2. A beggar wrapped himself with a few layers of newspaper on a cold winter night. This helped him to keep himself warm because (a) friction between the layers of newspaper produces heat. (b) air

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trapped between the layers of newspaper is a bad conductor of heat.

NCERT Exemplar solutions for Class 7 Science Chapter 4 - Heat

Throw out damaged or outdated test strips. Store strips in their sealed container; keep them away from heat, moisture and humidity. Be sure the strips are meant for your specific

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glucose meter.

Extreme temperatures:

Keep your glucose meter and test strips at room temperature.

Alcohol, dirt or other substances on your skin

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