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Element Atomic radius (nm) Ionic radius (nm)
W 0.114 0.195 X 0.072 0.136 Y 0.133 0.216
Z 0.099 0.181 (a) Would these form part of a metallic or a non-metallic group? Explain

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(26-50) 26) The largest number of molecules is in (a) 36 g of water (b) 28 g of carbon monoxide (c) 46 g of ethyl alcohol (d) 54 g of nitrogen pentoxide

Answer: A. 27) What are isobars?

Answer: Same mass number but a different atomic number.

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Chemistry. Form 2. The Structure of the Atom and the Periodic Table. 1.1 The structure of the atom. When scientists started exploring matter, they realised that matter can be divided into smaller and still smaller particles. They called the smallest particle an 'atom'. The

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name 'atom' was derived from the Greek word 'atomos', meaning ...

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INSTRUCTIONS Attempt all the question
in the question in the spaces provided
All the working must be clearly shown

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where appropriate. Negligence will be penalized. QUESTIONS 1. The table below shows some information concerning partials of certain elements. Element
Atomic number Number of electrons
Mass Number P 19 18 39

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FORM TWO NATIONAL ASSESSMENT 032
CHEMISTRY Time: 2:30 Hours Thursday,
14 t h November 2019 a.m. Instructions
1. This paper consists of sections A and
B with a total of ten (10) questions. 2.
Answer all questions in the spaces
provided 3. All writing must be in black
or blue ink except diagrams which must

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NaX. 1.00 g of this compound was

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dissolved in water to form a 100.0 mL solution.

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Boy (8-10) in Chemistry Lab. Charles Thatcher, Getty Images If there had been a million dollars on the line for knowing all the answers, you would have

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just missed the prize. It's OK to miss a question or two. That just shows you have other interests! You have a good command of chemistry.

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TO CANDIDATES 1. Answer all the
questions in the spaces provided. 2.
Mathematical tables and electronic
calculators may be used. 3. All working

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must be clearly shown where necessary
1. The table below shows the pH values
of solutions J to N.

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